

The $[\text{U}_2\text{I}_{10}]^{2-}$ Anion: Synthesis and Structure of $[\text{Ta}_7(\text{Se}_2)_{14}][\text{U}_2\text{I}_{10}]_2$

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Abstract. The new anion $[\text{U}_2\text{I}_{10}]^{2-}$ has been isolated in the compound $[\text{Ta}_7(\text{Se}_2)_{14}][\text{U}_2\text{I}_{10}]_2$. This compound was synthesized from the elements at 1173 K. It crystallizes in space group *Cccm* of the orthorhombic system with four formula units in a cell of dimensions $a = 21.5134(9)$ Å, $b = 22.1619(9)$ Å, and $c = 16.7834(6)$ Å. The crystal structure comprises infinite $[\text{Ta}(\text{Se}_2)_2]_n$ cationic chains charge balanced

by $[\text{U}_2\text{I}_{10}]^{2-}$ anions. The new $[\text{U}_2\text{I}_{10}]^{2-}$ anion is formed by two edge-sharing UI_6 octahedra. The U–I bond lengths are normal for tetravalent uranium. For charge balance, the $[\text{Ta}_7(\text{Se}_2)_{14}]^{4+}$ cation must contain three Ta^{IV} and four Ta^{V} atoms. Charge is delocalized over this cation as the Ta··Ta interactions differ minimally.

Introduction

Although some anions of formula $[\text{M}_2\text{X}_{10}]^{n-}$ ($M = \text{metal}$; $X = \text{F, Cl, Br, I}$) are known, there are none known for $M = \text{U}$ and only two confirmed crystallographically for $X = \text{I}$, namely $[\text{Sb}_2\text{I}_{10}]^{4-}$ [1] and $[\text{Bi}_2\text{I}_{10}]^{4-}$ [2, 3]. It is well known that metal complexes can be stabilized by large counterions [4]. Consequently, the discovery of a new, large anion, namely $[\text{U}_2\text{I}_{10}]^{2-}$, is of significance. The synthesis and structure of the compound $[\text{Ta}_7(\text{Se}_2)_{14}][\text{U}_2\text{I}_{10}]_2$ are described here.

Experimental Section

Black crystals of $[\text{Ta}_7(\text{Se}_2)_{14}][\text{U}_2\text{I}_{10}]_2$ were prepared from tantalum (Aldrich, 99.9%), selenium (Cerac, 99.99%), ^{238}U (ORNL), and iodine at 1173 K. A mixture of uranium (0.168 mmol), tantalum (0.336 mmol), iodine (0.504 mmol), and selenium (3.865 mmol) was loaded into a fused-silica tube. Afterwards, the tube was evacuated and flame-sealed. The reaction tube was heated in a computer-controlled furnace to 513 K in 2 h, to 713 K in 99 h, and afterwards to 1173 K in 46 h. Finally, it was slowly cooled at $5 \text{ K}\cdot\text{h}^{-1}$ to 373 K and crash cooled to 293 K. The presence of all four elements was confirmed by EDS analysis. The material is air sensitive.

Single-crystal X-ray diffraction data were collected with the use of graphite-monochromatized Mo- K_α radiation ($\lambda = 0.71073$ Å) at 100 K with a Bruker APEXII CCD diffractometer [5]. The crystal-to-detector distance was 5.023 cm. Crystal decay was monitored by recollecting 50 initial frames at the end of data collection. Data were collected by a scan of 0.3° in ω in groups of 606 frames at φ settings of 0° , 90° , 180° , and 270° . The exposure time was 10 s per frame. The collection of the intensity data and cell refinement were carried out with the program APEX2 [5]. Data reduction was carried out by using the pro-

gram SAINT+ [6]. Face-indexed absorption, incident beam, and decay corrections were performed numerically by using the program SADABS [6]. The structure was solved with the direct-methods program SHELXS and refined with the least-squares program SHELXL [7]. The program STRUCTURE TIDY [8] was then employed to standardize the atomic coordinates. Further details are given in Table 1 and in the cif file, which has been deposited [9].

Table 1. Crystal data and structural refinement for $[\text{Ta}_7(\text{Se}_2)_{14}][\text{U}_2\text{I}_{10}]_2$.

Compound	$[\text{Ta}_7(\text{Se}_2)_{14}][\text{U}_2\text{I}_{10}]_2$
Formula	$\text{I}_{20}\text{Se}_{28}\text{Ta}_7\text{U}_4$
$M / \text{g}\cdot\text{mol}^{-1}$	6967.65
Crystal system	orthorhombic
Space Group	<i>Cccm</i>
$a / \text{Å}$	21.5134 (9)
$b / \text{Å}$	22.1619 (9)
$c / \text{Å}$	16.7834 (6)
$V / \text{Å}^3$	8002.0 (5)
Temperature /K	100 (2)
Z	4
$\rho_c / \text{g}\cdot\text{cm}^{-3}$	5.784
μ (Mo- K_α) / cm^{-1}	380.55
Crystal size /mm	$0.36 \times 0.21 \times 0.10$
max/min transmission	0.1138/0.0255
$R(F)$ [$I > 2\sigma(I)$]	0.029
$R_w(F_o^{2a})$	0.069

$$a) R_w(F_o^2) = \{\sum [w(F_o^2 - F_c^2)^2] / \sum wF_o^4\}^{1/2}$$

Results and Discussion

The structure of $[\text{Ta}_7(\text{Se}_2)_{14}][\text{U}_2\text{I}_{10}]_2$ can be viewed as a packing of $[\text{Ta}_7(\text{Se}_2)_{14}]^{4+}$ chains and the $[\text{U}_2\text{I}_{10}]^{2-}$ anions (Figure 1). The chains extend along [010] and with the $[\text{U}_2\text{I}_{10}]^{2-}$ anions form alternating layers along [100]. There are two crystallographically independent $[\text{U}_2\text{I}_{10}]^{2-}$ anions in the asymmetric unit, each with crystallographically imposed $2/m$ symmetry. Each is formed by two edge-sharing UI_6 octahedra (Figure 2). Even though the structure of $A_2\text{UX}_5$ ($A = \text{K}$ and

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Rb, X = Cl, Br, and I) contains edge-sharing monocapped trigonal prisms that form chains [10] and the UI_6 species is well known in organometallic chemistry [11, 12], the $[U_2I_{10}]^{2-}$ anion does not appear to have been reported previously. The U–I bond lengths (Figure 2), which range from 2.9026(9) to 3.1270(5) Å, are normal for U^{IV} . For example, the U–I bond length in $Li_2[UI_6]$ is 3.013 Å [13]. The U(1)–U(1) and U(2)–U(2) distances are 4.719 and 4.653 Å, respectively, beyond the Hill limit for direct f–f orbital overlap [14], but slightly shorter than in $K_2[UI_5]$ where low temperature antiferromagnetic ordering was found [10].

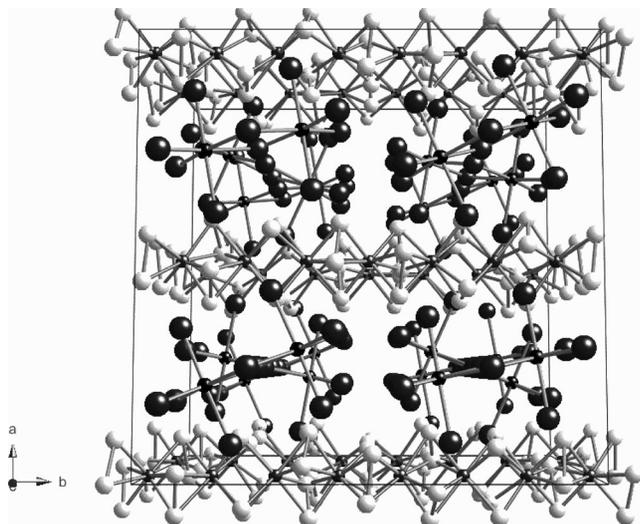


Figure 1. View approximately down [001] of the structure of $[Ta_7(Se_2)_{14}][U_2I_{10}]_2$. The cationic chains (small dark gray balls, tantalum; light large gray balls, selenium) extend along [010]. In the anions the small black balls are uranium, the large black balls are iodine.

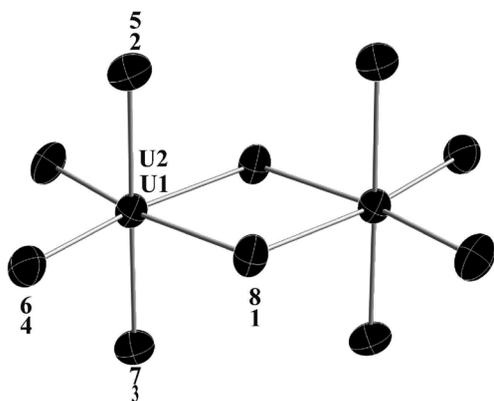


Figure 2. View of the $[U_2I_{10}]^{2-}$ anion in $[Ta_7(Se_2)_{14}][U_2I_{10}]_2$. Each of the crystallographically independent anions has imposed symmetry $2/m$. Distances: U(1)–I(1), 3.1270(5); U(2)–I(8), 3.1108(6); U(1)–I(2), 3.0000(9); U(2)–I(5), 3.0351(8); U(1)–I(3), 2.9039(9); U(2)–I(7), 2.9026(9); U(1)–I(4), 2.9406(6), U(2)–I(6), 2.9307(6) Å.

The $[Ta_7(Se_2)_{14}]^{4+}$ cation (Figure 3) comprises another example of Ta/Q_2 ($Q = S, Se,$ and Te) chains [15–19]. This anion has crystallographically imposed 222 symmetry at atom Ta(4). Each of the four crystallographically independent tanta-

lum atoms is coordinated by four Se_2 groups, the centers of which form a distorted tetrahedron. The Ta–Se distances ranging from 2.5685(8) to 2.7344(8) Å are normal as are the Se–Se single-bond lengths of 2.356(1) to 2.361(1) Å. For charge balance, the $[Ta_7(Se_2)_{14}]^{4+}$ cation must contain three Ta^{IV} and four Ta^V atoms. Charge is delocalized over this cation as the four $Ta \cdots Ta$ interactions of 3.1521(8) to 3.1760(4) Å differ minimally.

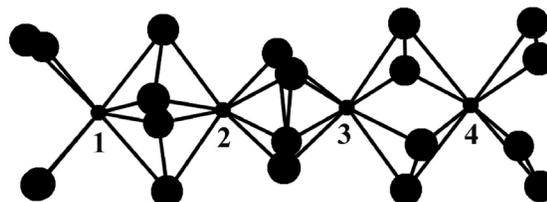


Figure 3. Part of the $[Ta_7(Se_2)_{14}]^{4+}$ anion in $[Ta_7(Se_2)_{14}][U_2I_{10}]_2$ with the tantalum atoms numbered. The rest of the anion is generated by a 222 symmetry operation that is at atom Ta(4). Distances: Ta(1)–Ta(1), 3.1521(8); Ta(1)–Ta(2), 3.1559(6); Ta(2)–Ta(3), 3.1730(6); Ta(3)–Ta(4), 3.1760(4); Ta–Se, 2.5685(8) to 2.7344(8); Se–Se, 2.356(1) to 2.361(1) Å.

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