Quadruple perovskites L'L''Ba₂Cu₂Ti₂O₁₁ as good candidates for high-temperature superconductors: Role of oxygen defects

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The electronic structure and properties of LaYBa $_2$ Cu $_2$ Ti $_2$ O $_{11}$, as a typical member of the L'L''Ba $_2$ Cu $_2$ Ti $_2$ O $_{11}$ family, were determined by means of the full-potential linear muffin-tin method. From the results obtained it appears that L'L''Ba $_2$ Cu $_2$ Ti $_2$ O $_{11}$ may be a good candidate for high- T_c superconductivity since its band structure at the Fermi energy is dominated by a quasi-two-dimensional Cu-O $dp\sigma$ band with a prominent van Hove singularity. We suggest that hole doping of about 0.6 holes per unit cell (or 0.3 holes per Cu-O layer) through mono- and/or divalent metal substitution for lanthanides may lead to an optimum T_c in these materials. The role of different structural imperfections on the T_c that apparently could be created in the process of chemical doping — oxygen impurities between the Cu-O layers and oxygen vacancies in the Ti-O plane — was also determined. We found that these likely structural imperfections are unfavorable for the creation of the superconducting state in these materials.

In the search for new and better high-temperature superconductors, many new families of layered perovskite structures have been discovered. One of them is the family of the quadruple perovskites $L'L''Ba_2Cu_2Ti_2O_{11}$ (L= lanthanide, Y). Gormezano and Weller^{1,2} discovered the layered compound Gd₂Ba₂Ti₂Cu₂O₁₁; subsequently, other independent investigations have shown that substitution of other lanthanides for gadolinium was possible.^{3–7} Recently, a comprehensive study of the crystal chemistry of L₂Ba₂Cu₂Ti₂O₁₁ (L = lanthanides) was carried out and in addition, layered materials with mixed lanthanide stoichiometries L'L"Ba₂Cu₂Ti₂O₁₁ were reported. These materials satisfy some general conditions for the occurrence of hightemperature superconductivity: they contain complete CuO₂ planes and a "charge reservoir" consisting of a double perovskite structure composed of L', Ti, and O. Thus, they may have good promise to be superconductors if appropriate doping of CuO2 planes is achieved. However, since recent attempts to substitute Gd by Ca in Gd₂Ba₂Ti₂Cu₂O₁₁ (Ref. 9) have not been successful, the question of whether it is possible to make these materials superconducting still remains open.

This paper addresses this question from the point of view of the electronic structure of one of those materials—LaYBa₂Cu₂Ti₂O₁₁—studied by means of the local-density approximation full-potential linear muffin-tin orbital method (FLMTO). Starting from the band-structure results for ideal (i.e., defect free) LaYBa₂Cu₂Ti₂O₁₁, we investigate the role of several possible structural imperfections — interlayer oxygen impurities (*p* type) and oxygen defects (*n* type) in the "charge reservoir."

The average structure of LaYBa₂Cu₂Ti₂O₁₁ (Ref. 8) is tetragonal with the lattice parameters a = 3.8969 Å and

c=15.716 Å and belongs to a P4/mmm space group, cf. Fig. 1. Atomic positions are listed in Table I. It was found that Y almost exclusively occupies the eight-coordinate site between Cu-O planes. We label the oxygen positions in the following way: O(1) is located in La-O plane, O(2) in the Ti-O plane, O(3) is an "apical" oxygen with respect to Cu-O planes and O(4) together with Cu atoms form the Cu-O layers. We have performed calculations for the defect-free LaYBa₂Cu₂Ti₂O₁₁ structure, for the structure containing oxygen between the Cu-O planes and for a structure containing 50% of O(2) vacancies. The calculations were performed using the FLMTO method¹⁰⁻¹³ in a spin-restricted scalarrelativistic mode with atomic Cu₃d¹⁰4s¹4p⁰, Ti $3d^24s^24p^0$, Y $4d^15s^25p^0$, Ba $6s^25p^65d^0$, La $6s^25p^65d^1$,

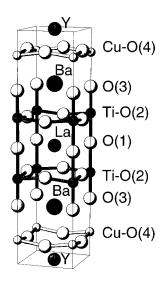


FIG. 1. Crystal structure of LaY Ba₂Cu₂Ti₂O₁₁.

TABLE I. Atomic coordinates (in units of the lattice constant) and muffin-tin (MT)-sphere radii for ideal LaYBa₂Cu₂Ti₂O₁₁ used in the FLMTO calculations (a=7.3636 a.u., c/a=4.0331) (Ref. 8).

Atom	x	у	z	R_{MT} , a.u.
Y	0.5	0.5	0.0	2.4
Ba	0.5	0.5	0.94	3.3
La	0.5	0.5	2.02	3.1
Cu	0	0	0.42	1.8
Ti	0	0	1.49	1.7
O(1)	0	0	2.02	1.9
O(2)	0.0	0.5	1.54	1.9
O(3)	0.0	0.0	1.02	1.64
O(4)	0	0.5	0.35	1.9

and O $2s^22p^4$ orbitals treated as valence band electrons and Cu $3s^23p^6$, Y $4p^6$, Ba $5p^6$ and Ti $3s^23p^6$ treated as semicore states (in a second energy window). We used a triple- κ basis set for each type of atom with angular momentum l up to 3 for Y, Ba, and La and up to 2 for the other atoms for $\kappa^2 = -0.01$ Ry, and with l up to 1 for $\kappa^2 = -1.0$ and -2.3 Ry. The charge density was calculated exactly in the muffin-tin spheres for angular momentum components up to l=5. The same l cutoff was used when interpolating in the interstitial region over Hankel functions with energies -1 and -3 Ry. The Brillouin-zone (BZ) integrations were carried out using an 84 k-point mesh.

In our discussion we will analyze the electronic structure of the ideal LaYBa₂Cu₂Ti₂O₁₁ and its changes related to crystal imperfections keeping in mind the common features of the electronic band structures of the known high- T_c superconductors. The most important seems to be the presence of a strongly hybridized Cu 3d-O 2p derived band crossing the Fermi energy (E_F), and the presence of a van Hove saddlepoint singularity (vHs) in the close vicinity of E_F , ^{14,15} which also appears to be favorable for superconductivity.

The FLMTO band structure of the *ideal* $LaYBa_2Cu_2Ti_2O_{11}$ in the vicinity of E_F along some symmetry lines in the simple tetragonal BZ is presented

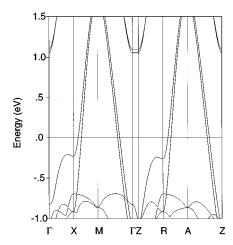


FIG. 2. FLMTO band structure of the ideal $LaYBa_2Cu_2Ti_2O_{11}$ structure in the Fermi energy region along some high-symmetry directions.

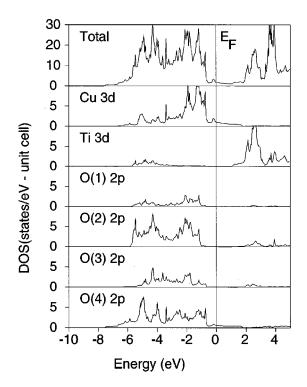


FIG. 3. Total and l- projected DOS for ideal LaYBa₂Cu₂Ti₂O₁₁.

in Fig. 2, where $X = (\pi/a;0;0)$, $M = (\pi/a;\pi/a;0)$, $R = (\pi/a;0;\pi/c)$, and $A = (\pi/a;\pi/a;\pi/c)$. Surprisingly, the band structure in the vicinity of E_F is very simple and looks like a band structure typical to all Cu-O based high- T_c superconductors. That is, only the quasi-two-dimensional Cu-O derived free-electron-like $dp\sigma$ bands cross E_F . This feature of the band structure suggests by analogy with other Cu-O based high- T_c superconductors that the undoped material is expected to be a Mott-Hubbard insulator. Indeed, the temperature dependence of the electrical resistivity of isostructural (and isoelectronic) materials exibit semiconductor behavior. The total and l-projected densities of states (DOS) for LaYBa₂Cu₂Ti₂O₁₁ are presented in Fig. 3. Again, as for all Cu-O based high- T_c superconductors, the only con-

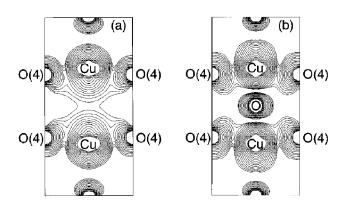


FIG. 4. Square of the wave function at the *X* point in the BZ normalized to two electrons per unit cell in the (010) plane for (a) ideal LaYBa₂Cu₂Ti₂O₁₁ and (b) LaYBa₂Cu₂Ti₂O₁₁ containing additional oxygen (O) between the Cu-O layers. Contours start from 10^{-3} e/a.u.³ and increase successively by a factor of $\sqrt{2}$.

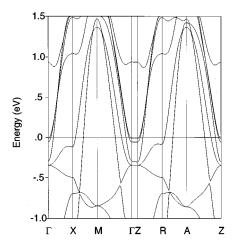


FIG. 5. FLMTO band structure in the Fermi energy region along some high-symmetry directions of the LaYBa₂Cu₂Ti₂O₁₁ containing additional oxygen between the Cu-O layers.

tribution to the DOS at the Fermi energy comes from the Cu d- and O(4) p- (in-plane oxygen) states. The prominent van Hove singularity is clearly seen on the DOS at 0.33 eV below E_F . The DOS composition at the E_F is also typical for high- T_c superconductors — it mainly consists of Cu 3d and O(4) 2p states. The two-dimensional character of the band can be clearly seen from the plot in the (010) plane of the square of the wave function at the X point in the BZ corresponding to the vHs [cf. Fig. 4(a)], which is mainly localized along the Cu-O planes. Thus, these features of the band structure we may expect LaYBa₂Cu₂Ti₂O₁₁, and most probably other members of the L'L"Ba₂Cu₂Ti₂O₁₁ family (since their crystal structure and valence electron concentration are practically the same) to display superconductivity, if properly doped. If we associate the "optimum doping" with the position of the Fermi energy in the close vicinity of the vHs, which seems to be quite relevant, 14,15 we may predict the hole doping to be about ~ 0.6 holes per unit cell, or ~ 0.3 holes per Cu-O layer. Most likely such a doping can be achieved by substituting mono- or divalent atoms for Y and/or lanthanides.

Next we investigated the effect of two possible crystal defects which can strongly interfere with effective doping and the "two dimensionality" of Cu-O $dp\sigma$ bond: (i) an oxygen impurity incorporated between the Cu-O layers along the Cu-Cu bond and (ii) the oxygen [O(2)] vacancy in Ti-O planes. Figure 5 shows the band structure of LaYBa₂Cu₂Ti₂O₁₁ containing additional oxygen between the Cu-O layers. In conjunction with Fig. 6 that presents the total and l-projected DOS, one can see that the main differences from the ideal LaYBa₂Cu₂Ti₂O₁₁ are (i) the lowering of the antibonding part of the Ti 3d states down to E_F because of the change in electrostatic potential of the Cu-O layers and (ii) the appearance of new interlayer oxygen states at E_F . These oxygen states contribute heavily to the former Cu-O $pd\sigma$ band and change its character. Figure 4(b) presents the squared wave function at the same X point in the BZ as for the ideal LaYBa₂Cu₂Ti₂O₁₁. Although the states of the interlayer oxygen at this point in the BZ do not form any chemical bonding with states of the Cu-O layers, they con-

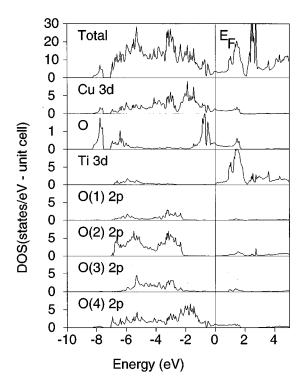


FIG. 6. Total and *l*-projected DOS for LaYBa₂Cu₂Ti₂O₁₁ containing additional oxygen between the Cu-O layers.

tribute strongly to the band composition. It is interesting to note that this one extra oxygen per unit cell, which one can think would heavily dope the Cu-O $pd\sigma$ band by holes, does not significantly change the energy difference between the vHs and E_F . This suggests that the extra oxygen in between the Cu-O layers is mostly in its atomiclike state. Consequently, any doping coming from the "charge reservoir" would most probably first localize on that interlayer oxygen and prevent the $pd\sigma$ band from being optimally doped. Also, the incorporation of this extra oxygen apparently changes the two-dimensional character of the $pd\sigma$ band, which is thought to be responsible for the high- T_c properties of Cu-O layered ceramics. Thus, we can speculate on why the attempts to make L'L"Ba2Cu2Ti2O11 superconducting have failed so far. Most probably, in the process of doping some additional oxygen is incorporated between the Cu-O layers and this alters the character of the states at the E_F and destroys the main condition for high- T_c superconductivity.

Finally, we examined the electronic structure of LaY $Ba_2Cu_2Ti_2O_{11}$ containing 50% oxygen vacancies at the O(2) site. The band structure plotted in Fig. 7 shows that additional bands appear in the vicinity of E_F . From the total and l-decomposed DOS (Fig. 8), one can see that Ti d states shift down in energy and now contribute to the DOS at E_F . The nature of those states can be understood from their specific space localization. A detailed analysis of the charge distribution coming from these states shows that they are mainly localized in the region of the Ti-O plane.

Figure 9 shows the contour map of the square of the wave function associated with the topmost band below E_F at the X point in the BZ. The appearance of such "vacancy states" below E_F is caused by rearrangement of electron states around oxygen vacancies with a decisive role played by changes of wave functions of Ti atoms closest to the va-

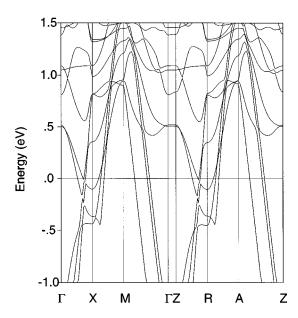


FIG. 7. FLMTO band structure in the Fermi energy region along some high-symmetry directions of the LaY $Ba_2Cu_2Ti_2O_{11}$ containing oxygen vacancies in the Ti-O layers.

cancy. Specifically, the appearance of an O(2) vacancy leads to the breakdown of direct Ti-O bonding and to the creation new Ti-Ti bonds through the vacancy region. This also was found to be a common feature of the electronic structure of carbides and nitrides of transition metals containing metalloid vacancies. ¹⁶ Such an analysis suggests that "extra" electrons created in the Ti-O layers by oxygen vacancies will not

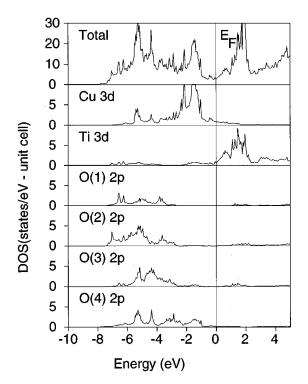


FIG. 8. Total and *l*-projected DOS for LaYBa₂Cu₂Ti₂O₁₁ containing oxygen vacancies in the Ti-O layers.

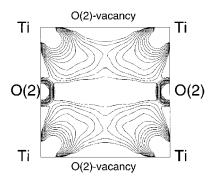


FIG. 9. Square of the wave function at the X point in the BZ normalized to two electrons per unit cell in the (001) Ti-O plane for LaYBa₂Cu₂Ti₂O₁₁ containing oxygen vacancies in the Ti-O layers. Contours start from 10^{-3} e/a.u.³ and increase successively by a factor of $\sqrt{2}$.

be able to move to the Cu-O layers and change the carrier concentration since they stay in the vicinity of O vacancies taking part in Ti-Ti bonding. It is also interesting to find the character of the electron states associated with Cu-O planes that lie close to E_F . At the X point in the BZ, the band closest to E_F of Cu-O(4) character was found at ~ 0.5 eV below E_F . The square of the wave function of this band, presented in Fig. 10, is seen to differ dramatically from the character of the states in ideal LaYBa₂Cu₂Ti₂O₁₁ [Fig. 4(a)]. Thus, we should conclude that oxygen vacancies in the Ti-O plane are also not good for superconductivity since they (i) do not contribute to the doping ability of the "charge reservoir" and (ii) destroy the two-dimensional character of the Cu-O states at the Fermi energy.

To summarize, from FLMTO band structure calculations of LaYBa₂Cu₂Ti₂O₁₁ as a typical member of the L'L''Ba₂Cu₂Ti₂O₁₁ family we found, from the point of view of electronic structure, that the whole L'L''Ba₂Cu₂Ti₂O₁₁ family may be a good candidate for high- T_c superconductivity. Their band structure at E_F is dominated by a quasi-two-dimensional Cu-O $dp\sigma$ band with a prominent vHs. We suggest that hole doping of about 0.6 holes per unit cell (or 0.3 holes per Cu-O layer) through mono- and/or divalent metal

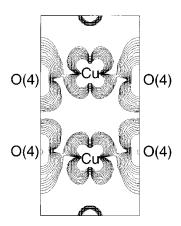


FIG. 10. Square of the wave function at the *X* point in the BZ normalized to two electrons per unit cell in the (010) Cu-O plane for LaYBa₂Cu₂Ti₂O₁₁ containing oxygen vacancies in Ti-O layers. Contours start from 10^{-3} e/a.u.³ and increase successively by a factor of $\sqrt{2}$.

substitution for lanthanides should lead to an optimum T_c in these materials. We have also investigated the role of different structural imperfections that are apparently created in the process of chemical doping (namely oxygen impurities between the Cu-O layers and oxygen vacancies in the Ti-O plane), and we have shown that they are unfavorable for the creation of superconducting state in these materials.

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